

Start



(Q) What does the voltage of a cell depend on?

- A - The size of the cell
- B - How much oxygen is present
- C - The difference in reactivity of the two electrodes

(A) The voltage of a cell depends on the difference in reactivity of the two electrodes.



(Q) Where can hydrogen be obtained from on the space station without transporting it from Earth?

- A - From plants
- B - From water
- C - From the moon

(A) Water can be electrolysed using electricity from solar cells. This produces hydrogen and oxygen.



(Q) Chemical cells produce a voltage. When do they stop producing a voltage and 'go flat'?

- A - When one of the reactants has run out.
- B - When they get too hot
- C - When they reach the expiry date on the cell

(A) A cell will produce a voltage until one of the reactants runs out.



(Q) Hydrogen can be used as a fuel for cars. Which of the following is an advantage of using hydrogen instead of petrol?

- A - It is cheaper
- B - Exhaust gases contain only water
- C - It is easier to store

(A) Hydrogen powered cars will produce only water in their exhaust gases.



(Q) What is the only chemical product when hydrogen and oxygen are reacted together in a fuel cell?





- A - Water
- B - Electricity
- C - Carbon Dioxide

(A) Water is the only chemical product from a hydrogen-oxygen fuel cell.



(Q) Which of the following is a disadvantage of using hydrogen as a fuel?

- A - There are currently very few hydrogen filling stations in the UK
- B - It directly contributes to global warming
- C - Hydrogen is radioactive

<p>(A) At present, there are very few hydrogen filling stations in the UK, but this will improve over time.</p> 	<p>(Q) Which is not an advantage for using a hydrogen-oxygen fuel cell in a space station?</p> <p>A - No moving parts B - Small and compact size C - They generate useful heat for warmth</p>
<p>(A) Fuel cells do not generate much heat due to their efficiency, so they are not useful for keeping the space station warm.</p> 	<p>(Q) What is a disadvantage of a rechargeable cell when compared with an alkaline cell?</p> <p>A - It is smaller B - It costs more to manufacture C - It can be reused many times.</p>
<p>(A) Rechargeable cells cost more to manufacture than alkaline cell.</p> 	<p>(Q) In addition to a fuel, what other chemical is needed by a fuel cell?</p> <p>A - Nitrogen B - Oxygen C - Methane</p>
<p>(A) Fuel cells all require oxygen. Often this comes from the air.</p> 	<p>(Q) In the space station, why can you not rely on solar cells all the time?</p> <p>A - They have to be shut down for maintenance B - They are unreliable and frequently overheat C - They only produce electricity when the sun is visible</p>
<p>(A) Solar cells only work when sunlight falls on them, and this is not always the case on the space station.</p>	<p style="text-align: center; font-size: 2em;">End</p> 